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## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.005 \AA$
$R$ factor $=0.050$
$w R$ factor $=0.154$
Data-to-parameter ratio $=9.2$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## 16a,17a-Epoxy-20-oxopregn-5-ene-3/,21-diyl diacetate

The title compound, $\mathrm{C}_{25} \mathrm{H}_{34} \mathrm{O}_{6}$, has a $3 \beta, 21$-configuration, with the epoxy O atom at $16 \alpha, 17 \alpha$. Rings $A$ and $C$ have chair conformations which are slightly flattened. Because of the presence of a double bond, ring $B$ assumes an $8 \beta, 9 \alpha$-half chair conformation. Ring $D$ has a conformation close to a $14 \alpha$ envelope. Cohesion of the crystal can be attributed to van der Waals interactions and weak $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions.

## Comment

Epoxides are very reactive compounds and, owing to the possibility of performing ring opening both regio- and stereospecifically, they are used as versatile synthetic intermediates in the preparation of complex bioactive molecules. Therefore, our interest in preparing new steroids for biological screening (Moreno et al., 1998; Campos Neves et al. 1999) led us to the title compound, (I), a key intermediate in the bisoxidation, at atoms C 16 and C 21 , of 20 -oxopregnenes.

(I)

Compound (I) was obtained from 20-oxopregne-5,16-dien$3 \beta$-yl acetate in a four-stage synthesis: stereoselective epoxidation of the C16 double bond (Moreno et al., 1998); careful oxidation at C 21 with $\mathrm{PhI}(\mathrm{OAc})_{2}$ (Kamernitzky et al., 1985); subsequent hydrolysis of the corresponding 21-hydroxyketone dimethyl acetal (Kamernitzky et al., 1985) and acetylation of the $3 \beta-\mathrm{OH}$ and $21-\mathrm{OH}$. According to the X-ray analysis of the title compound, (I), all rings are trans fused. An ORTEPII (Johnson, 1976) drawing of the molecule with the atomic numbering scheme is shown in Fig. 1. The distance between the terminal O atoms, $\mathrm{O} 22 \cdots \mathrm{O} 24$, is $13.761(6) \AA$, and between the terminal C atoms, $\mathrm{C} 23 \cdots \mathrm{C} 25$, is 17.363 (7) $\AA$. Bond lengths and angles are within the expected ranges (Allen et al., 1987), with average values $\mathrm{Csp}{ }^{3}-\mathrm{Csp}^{3}=1.528$ (13), $\mathrm{Csp}{ }^{3}-\mathrm{Csp} p^{2} 1.497$ (15), $\mathrm{Csp}^{2}=\mathrm{Csp}^{2} 1.332$ (4) and $\mathrm{C}=\mathrm{O}$ 1.209 (15) $\AA$. It is worth mentioning a small but significant asymmetry between the two $\mathrm{C}=\mathrm{O}$ acetoxy bond lengths [1.232 (6) and 1.191 (6) $\AA$ ] and the two C-O epoxy bond lengths [1.424 (5) and 1.461 (4) Å]. Similar asymmetries were also found in the $3 \beta$-yl acetate of this compound (Andrade et al., 2001). Rings $A$ and $C$ are slightly flattened, the mean values of their torsion angles being 51.8 (2) and $55.9(1)^{\circ}$,

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Figure 1
The molecular structure of (I), showing 50\% displacement ellipsoids.
respectively. The presence of the acetoxy group bonded to C3 does not disturb the usual chair conformation of ring $A$ the of steroid nucleus. Due to the $\mathrm{C} 5=\mathrm{C} 6$ double bond, the environment of atom C5 is planar (the sum of the valence angles around this atom is $359.9^{\circ}$ ) and consequently ring $B$ is distorted, assuming an $8 \beta, 9 \alpha$-half-chair conformation with asymmetry parameters (Duax \& Norton, 1975) $\Delta C_{2}(5,6)=$ $3.1(4)^{\circ}, \Delta C_{2}(5,10)=48.6(4)^{\circ}$ and $\Delta C_{s}(6)=20.0$ (3). The fivemembered ring $D$ assumes a $14 \alpha$-envelope conformation, with puckering parameters (Cremer \& Pople, 1975) $q_{2}=$ 0.388 (4) $\AA$ and $\varphi_{2}=213.6$ (6) ${ }^{\circ}$ [pseudo-rotation (Altona et al., 1968) and asymmetry parameters: $\Delta=-31.4$ (7), $\tau_{m}=39.5$ (2), $\Delta C_{s}(14)=1.8(4)$ and $\left.\Delta C_{2}(13,14)=16.9(4)^{\circ}\right]$. The $3 \beta$-acetoxy group attached to ring $A$ is planar. The $\mathrm{C} 3-\mathrm{O} 3$ bond is oriented equatorially. The dihedral angle between the plane of the $3 \beta$-acetoxy group and the mean molecular plane is $61.3(1)^{\circ}$. The dihedral angle between the plane defined by the epoxy group and the average molecular plane comprising rings $A, B, C$ and $D$ is $87.8(2)^{\circ}$. As reported for a similar structure (Andrade et al., 2001), the epoxide link on ring $D$ may be responsible for the unusual conformation of the substituent group at C 17 , where the $\mathrm{C} 13-\mathrm{C} 17$ bond almost eclipses the $\mathrm{C} 20-\mathrm{O} 20$ bond $[\mathrm{C} 13-\mathrm{C} 17-\mathrm{C} 20-\mathrm{O} 20$ $\left.-5.8(6)^{\circ}\right]$. This conformation might be responsible for the relatively large value of the pseudo-torsion angle $\mathrm{C} 18-\mathrm{C} 13-$ $\mathrm{C} 10-\mathrm{C} 1912.4$ (2) ${ }^{\circ}$. The $17 \beta$ group may be divided into an acetyl group and an acetoxy group. The dihedral angles between the planes of the acetyl group and the acetoxy group with the average molecular plane comprising rings $A, B, C$ and $D$ are 28.9 (2) and $59.0(2)^{\circ}$, respectively. The environments of atoms C20 and C22 are planar (the sums of the valence angles around these C atoms are both $360.0^{\circ}$ ). However, the angle between the plane of the acetyl group and the plane of the acetoxy group, $82.5(2)^{\circ}$, shows a twist of the two groups around the $\mathrm{C} 21-\mathrm{O} 21$ bond.

The molecule lacks a strong hydrogen donor, so no classical hydrogen bonds are present in the structure and its cohesion is mainly achieved by van der Waals interactions. One intramolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ short contact between the O atom of the $17 \beta$-group and a neighbouring H atom of a methyl group is present, $\mathrm{C} 18-\mathrm{H} 18 \cdots \mathrm{O} 20$ at 3.107 (5) $\AA$. The bent angle
defined by these atoms, $116^{\circ}$, and the weak acidic character of the methyl group suggest that this interaction is probably destabilizing and should not be classified as a weak hydrogen bond. There is, in addition, an intermolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ short contact between two atoms of the $17 \beta$-group: $\mathrm{C} 16-$ $\mathrm{H} 16 \cdots \mathrm{O} 22^{\mathrm{i}}\left[\mathrm{C} 16 \cdots \mathrm{O} 22^{\mathrm{i}} 3.127\right.$ (6) $\AA$ and $\mathrm{C} 16-\mathrm{H} 16 \cdots \mathrm{O} 22^{\mathrm{i}}$ $115^{\circ}$; symmetry code: (i) $\left.1+x, y, z\right]$.

## Experimental

Stereoselective epoxidation of the C16 double bond of the commercially available 20 -oxopregne- 5,16 -dien- $3 \beta$-yl acetate, with hydrogen peroxide in alkaline conditions, was performed according to the literature (Moreno et al., 1998; Andrade et al., 2001). Crystals suitable for X-ray analysis were obtained by slow evaporation of a solution of the steroid in methanol/ethyl acetate.

## Crystal data

$\mathrm{C}_{25} \mathrm{H}_{34} \mathrm{O}_{6}$
$M_{r}=430.52$
Orthorhombic, $P 2_{1} 2_{1} 2_{1}$
$a=5.988$ (1) £
$b=10.668$ (1) $\AA$
$c=36.511$ (3) $\AA$
$V=2332.4(5) \AA^{3}$
$Z=4$
$D_{x}=1.226 \mathrm{Mg} \mathrm{m}^{-3}$

## Data collection

## Enraf-Nonius MACH-3

diffractometer
Profile data from $\omega-2 \theta$ scans
Absorption correction: $\psi$ scan
(North et al., 1968)
$T_{\text {min }}=0.887, T_{\text {max }}=0.913$
6910 measured reflections
2624 independent reflections
1949 reflections with $I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.050$
$w R\left(F^{2}\right)=0.154$
$S=1.03$
2624 reflections
284 parameters
H -atom parameters constrained
$\mathrm{Cu} K \alpha$ radiation
Cell parameters from 25 reflections
$\theta=18.8-26.5^{\circ}$
$\mu=0.70 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Prism, colourless
$0.49 \times 0.14 \times 0.13 \mathrm{~mm}$

$$
\begin{aligned}
& R_{\text {int }}=0.068 \\
& \theta_{\max }=72.6^{\circ} \\
& h=-7 \rightarrow 7 \\
& k=-13 \rightarrow 13 \\
& l=0 \rightarrow 45 \\
& 3 \text { standard reflections } \\
& \quad \text { frequency: } 200 \text { min } \\
& \quad \text { intensity decay: } 37.0 \%
\end{aligned}
$$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0947 P)^{2}\right. \\
& \quad+0.2153 P] \\
& \text { where } P=\left(F_{o}{ }^{2}+2 F_{c}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.005 \\
& \Delta \rho_{\max }=0.29 \mathrm{e}^{-3} \\
& \Delta \rho_{\min }=-0.19 \mathrm{e}^{-3}
\end{aligned}
$$

Intensities were corrected for the large (linear) decay of $37 \%$, which appears to be due to radiation damage. The H atoms were placed at calculated idealized positions and refined as riding on their parent atoms using SHELXL97 defaults. The absolute configuration of the molecule is known from the synthesis route. Due to the lack of significant anomalous scattering, the Friedel pairs were merged before refinement. Examination of the crystal structure with PLATON shows that the crystal structure does not contain voids which might be occupied by solvent molecules.

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: CAD-4 Software; data reduction: HELENA (Spek, 1997); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL97.

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